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Computational Modelling Study on the Stability Li_{1.2}Mn_{0.8}O₂ Cathode Material

Due to the rising demand of renewable energy, lithium-ion batteries have attracted much attention with Li₂MnO₃ being a perfect candidate to use as the cathode material. This is due to its high energy density and specific capacity. However, Li₂MnO₃ suffers from poor cycling stability and voltage fade which limits its practical application. In this work the built monoclinic Li_{1.2}Mn_{0.8}O₂ is doped with Ti to attain the fundamental understanding of the crystal cycling stability. With the application of first-principles calculations combined with the ground state search, this study will generate phases of the Ti doped Li_{1.2}Mn_{0.8}O₂ clusters. The ground state search was able to generate 20 new phases which are thermodynamically stable with negative enthalpy of formation. The cross-validation score was found to be less than 5 meV/atom which indicate accuracy of ground-state search calculations allowing the temperature profile to be implied on the different phases which showed that phase transition occurs at 1200K. The analysis paves the way for further research on the crystal's stability as a function of temperature.

Keywords: Li-ion batteries, Cathode, First-principles, Cluster Expansion, Cross-Validation Score, Monte-Carlo.

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Yes

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